

**CHEM 1304 General Inorganic Chemistry II  
Spring 2024**

<b>General Course Information</b>	
<b>Information Item</b>	<b>Information</b>
<b>Instructor:</b>	<i>Dr. Gina Chiarella</i>
<b>Section # and CRN:</b>	<i>Z02 - 24612</i>
<b>Office Location:</b>	<i>Room 230AF - E. E. O'Bannion</i>
<b>Office Phone:</b>	<i>936-261-3113</i>
<b>Email Address:</b>	<i><a href="mailto:gmchiarella@pvamu.edu">gmchiarella@pvamu.edu</a></i>
<b>Office Hours:</b>	<i>T, R from 12:00 to 3:00 pm (room 230AF or 221 or Online)</i>
<b>Mode of Instruction:</b>	<b>Online</b>
<b>Course Location:</b>	<b>Online</b>
<b>Response-time for feedback and grading of assignments and exams</b>	<b>Students can expect feedback and grades on assignments and exams <b>WITHIN ONE WEEK</b>, unless otherwise noted.</b>
<b>Catalog Description:</b>	Bonding theory and molecular structure, intermolecular forces properties of solutions, chemical kinetics, chemical equilibrium, acid-based equilibria, thermodynamics, electrochemistry and nuclear chemistry and introduction to organic chemistry.
<b>Prerequisites:</b>	MATH 1113 and CHEM 1303.
<b>Co-requisites:</b>	None
<b>Required Text(s):</b>	Chemistry & Principles and Reactions – Masterton and Hurley 8th edition
<b>Recommended Text(s):</b>	Chemistry an Atoms-Focused Approach, First Edition Thomas R. Gilbert, Rein V. Kirss, Natalie Foster, Geoffrey Davies. ISBN-13: 9780393912340

General Course Information Table

**Student Learning Outcomes:**

<i>Core Curriculum Outcomes *</i>
(A) <b>Critical Thinking Skills:</b> to include creative thinking, innovation, inquiry, and analysis, evaluation and synthesis of information;
(B) <b>Communication Skills:</b> to include effective development, interpretation and expression of ideas through written, oral and visual communication;
(C) <b>Empirical and Quantitative Skills:</b> to include the manipulation and analysis of numerical data or observable facts resulting in informed conclusions;
(D) <b>Teamwork:</b> to include the ability to consider different points of view and to work effectively with others to support a shared purpose or goal;
(E) <b>Personal Responsibility:</b> to include the ability to connect choices, actions and consequences to ethical decision-making; and
(F) <b>Social Responsibility:</b> to include intercultural competence, knowledge of civic responsibility, and the ability to engage effectively in regional, national, and global communities.

<b>Student Learning Outcomes</b>	
<b>SLO #</b>	<b>Upon successful completion of this course, students will be able to:</b>
SLO 1	Learn basic chemical concepts: mass, energy, chemical processes, chemical reactions, mol,

	molecules, atoms.
SLO 2	Analyze chemical problems and develop mathematical skill to resolve stoichiometric problems.
SLO 3	Recognize the scientific approach and methods involving making observations and gathering data.
SLO 4	Dominate the knowledge of the three Laws of Thermodynamics and perform energy balance calculations.
SLO 5	Gain a basic understanding of periodic properties of elements and chemical bonding.
SLO 6	Apply knowledge of chemistry to everyday life and explain the observation and changes.

Student Learning Outcomes Table

Unit # and Title							
Unit/Module-Level Objective	SLO1	SLO2	SLO3	SLO4	SLO5	SLO6	Materials, Activities & Assessments
CH 1: Rate of Reaction	x	x	x	x	x	x	Online homework
CH 2: Gaseous chemical Equilibrium	x	x	x	x	x	x	Online homework
CH 3: Acids and bases	x	x	x	x	x	x	Online homework
CH 4: Equilibria in acid-base	x	x	x	x	x	x	Online homework
CH 5: Complex ion and Precipitation equilibria	x	x	x	x	x	x	Online homework
CH 6: Spontaneity of Reaction	x	x	x	x	x	x	Online homework
CH 7: Electrochemistry	x	x	x	x	x	x	Online homework
CH 8: Nuclear Reactions	x	x	x	x	x	x	Online homework
CH 9: Complex Ions	x	x	x	x	x	x	Assignment
CH 10: Chemistry of the Metals	x	x	x	x	x	x	Assignment

## Major Course Requirements

### Method of Determining Final Course Grade

Course Grade Requirement	Value	Total
1. Assignments	4% (20 p)	4% (20 p)
2. Homework	16% (80 p)	16% (80 p)
3. Partial Exam (4) the lowest grade will be dropped	20% x 3 (100 p x 3)	60% (300 p)
4. Final exam	20% (100 p)	20% (100 p)
5. In-class exercises (Extra credits)	-----	----
<b>Total:</b>		<b>100% (500 p)</b>

Course Grade Requirement Table

### Grading Criteria and Conversion:

A = 90-100 % (450-500 p)

B = 80-99.9 % (400-449 p)

C = 70-79.9% (350-399 p)

D = 60-69.9 % (300-349 p)

F = below 60% (below 300 p)

### Detailed Description of Major Assignments

Assignment Title or Grade Requirement	Description
1. Homework + Assignment	2 assignments and 8 homework sets. All homework assignments are due at the Time posted on e-courses
2. Partial exams	4 partial exams, the lowest grade is dropped. Exams should be taken as scheduled.

	No makeup examinations will be allowed except under documented emergencies (See Student Handbook). A scientific calculator is required for exam taking. Students will not be allowed to use telephone or other communicating devices to make calculations
3. Final exam	Attendance to the final exam is a requirement to pass the course
4. In-class exercises	Will take place in determinate dates and will not be make-ups

Detailed Description of Major Assignments Table

<b>Course Procedures or Additional Instructor Policies</b>	
<b>Exams policy</b>	
All partial exams will be taken online for a period of one academic hour (50 minutes) in any of the exam periods on a Friday date.	
Exam periods are:	
From 4:00 to 4:50 pm	
From 6:00 to 6:50 pm	
From 8:00 to 8:50 pm	
From 10:00 to 10:50 pm	
<b>The final exam is a common exam and will be taken face-to-face along with all the other sections of this course.</b>	
Students to fail to take an exam may ask to take a makeup only by submission of a valid excuse and approval of this document.	
<b>Any makeup exam will be offered at a fixed time for all students whose excuse was accepted by the professor of this course.</b>	
<b>There are no makeups for exams already taken, independently of the grade attained and loss of Wi-Fi connection by the student's computer.</b>	
<b>Proctorio Configuration</b>	
Enable Proctorio Secure Exam Proctor; the configuration of Proctorio is attached to this syllabus (Attachment 1).	
<b>Short answers.</b>	
Most of the exams will contain short answers and numbers with powers of ten; the way to write them on canvas is attached (Attachment 2)	
<b>Your Wi-Fi connection and computer is your responsibility</b>	
<b>Taskstream</b>	
Taskstream is a tool that Prairie View A&M University uses for assessment purposes. One of your assignments may be required to be submitted as an "artifact," an item of coursework that serves as evidence that course objectives are met. If applicable, more information will be provided during the semester by your department, but for general information, you can visit Taskstream via the link in eCourses.	

### Semester Calendar

<b>Week One:</b>	<b>Chapter 1 - Rate of reaction</b>
<b>Topic Description</b>	
Readings:	M = Reaction Rate - Dependence on concentration of reactants. W = First-order reaction F = Reactions of Other Orders
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Two:</b>	<b>Chapter 1 - Rate of reaction</b>
<b>Topic Description</b>	
Readings:	M = Models for Reaction Rate: Collision model - Activated complex W = Effect of Temperature - Arrhenius Equation F = Catalysis - Reaction Mechanism
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Three:</b>	<b>Chapter 2 - Gaseous chemical Equilibrium</b>
<b>Topic Description</b>	
Readings:	M = The Equilibrium Constant, K - Only gases involved – Solids, liquids and gases W = Aqueous solutions

Assignment (s):	F = Relations between equilibrium constants M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Four:</b>	<b>Chapter 2 Gaseous - chemical Equilibrium</b>
Topic <b>Description</b>	
Readings:	M = Determination of direction of reaction W = Extent of reaction F = Effect of Changes in Conditions on an Equilibrium System
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Five:</b>	<b>Chapter 3 - Acids and bases</b>
Topic <b>Description</b>	
Readings:	<b>M = Exam I</b> W = Brønsted-Lowry Model - Acidic and Basic Water Solutions F = pH and pOH
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Six:</b>	<b>Chapter 3 - Acids and bases</b>
Topic <b>Description</b>	
Readings:	M = Weak Acids - Equilibrium constants W = Molecular structure and acid strength F = Acid-Base Properties of Salt Solutions - The Lewis Model
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Seven:</b>	<b>Chapter 4 - Equilibria in acid-base - Solutions</b>
Topic <b>Description</b>	
Readings:	M = Buffers - Preparation - Effect of adding a strong acid or base to a buffer W = Indicators - F = Acid-Base Titrations
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Eight:</b>	<b>Chapter 5 – Complex ion and Precipitation equilibria</b>
Topic <b>Description</b>	
Readings:	M = Complex Ion Equilibrium: $K_f$ W = Solubility Equilibrium: $K_{sp}$ F = Precipitate Formation - Dissolving Precipitates
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Nine:</b>	<b>Chapter 6 - Spontaneity of Reaction</b>
Topic <b>Description</b>	
Readings:	<b>M = Exam II</b> W = Spontaneous Reactions F = Factors affecting spontaneity
Assignment (s):	M = On-line Homework W = On-line Homework F = On-line Homework
<b>Week Ten:</b>	<b>Chapter 6 - Spontaneity of Reaction</b>
Topic <b>Description</b>	
Readings:	M = Entropy Changes W = Calculation of $\Delta S^\circ$ F = Reactions in which $\Delta S^\circ$ is positive
Assignment (s):	M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Eleven:**

**Chapter 6 - Spontaneity of Reaction**

Topic **Description**

Readings:

M = Free Energy Changes - Effect of  $\Delta H^\circ$  and  $\Delta S^\circ$  on spontaneity

W = Calculation of  $\Delta G^\circ$  from  $\Delta H^\circ$  and  $\Delta S^\circ$

F = Relationship between  $\Delta G^\circ$  and k

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Twelve:**

**Chapter 7 - Electrochemistry**

Topic **Description**

Readings:

M = Redox Reactions - Balancing Redox Equations - Voltaic Cells

W = Relative strengths of oxidizing and reducing agents

F = Calculation of  $E^\circ$

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Thirteen:**

**Chapter 7 - Electrochemistry**

Topic **Description**

Readings:

M = Relation among  $E^\circ$ ,  $\Delta G^\circ$  and K - Nernst equation -

**W = Exam III**

F = Electrolytic Cells

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Fourteen:**

**Chapter 8 - Nuclear Reactions**

Topic **Description**

Readings:

M = Nuclear Stability – Radioactivity

W = Rate of Decay - Mass Defect

F = Fission – Fusion

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Fifteen:**

**Chapter 9 - Complex Ions**

Topic **Description**

Readings:

M = Complex ions - Nature of ligands - Naming Coordination Compounds

W = Geometry - Electronic Structure of Transition Metal Cations

F = Crystal Field Model – Formation Constants

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Week Sixteen:**

**Chapter 10 - Chemistry of the Metals**

Topic **Description**

Readings:

M = Metallurgy of chlorides, oxides and sulfides - Reactions of Group 1, Group 2

W = Redox Chemistry of the Transition Metals

**F = Exam IV**

Assignment (s):

M = On-line Homework

W = On-line Homework

F = On-line Homework

**Final Exam**

**Student Support and Success**

## John B. Coleman Library

The John B. Coleman Library's mission is to enhance the scholarly pursuit of knowledge, to foster intellectual curiosity, and to promote life-long learning and research through our innovative services, resources, and cultural programs, which support the Prairie View A&M University's global mission of teaching, service, and research. It maintains library collections and access both on campus, online, and through local agreements to further the educational goals of students and faculty. [Library Website](#) Phone: 936-261-1500

## Academic Advising Services

Academic Advising Services offers students various services that contribute to student success and lead toward graduation. We assist students with understanding university policies and procedures that affect academic progress. We support the early alert program to help students connect to success early in the semester. We help refer students to the appropriate academic support services when they are unsure of the best resource for their needs. Faculty advisors support some students in their respective colleges. Your faculty advisor can be identified in PantherTracks. Advisors within Academic Advising Services are available to all students. We are located across campus. Find your advisor's location by academic major on the [advising website](#). Phone: 936-261-5911

## The University Tutoring Center

The University Tutoring Center (UTC) offers free tutoring and academic support to all registered PVAMU students. The mission of the UTC is to help provide a solid academic foundation that enables students to become confident, capable, independent learners. Competent and caring staff and peer tutors guide students in identifying, acquiring, and enhancing the knowledge, skills, and attitudes needed to reach their desired goals. Tutoring and academic support are offered face-to-face in the UTC and virtually in online sessions. Other support services available for students include Supplemental Instruction, Study Breaks, Academic Success Workshops, and Algebra Study Jam. Location: J. B. Coleman Library, Rm. 307; Phone: 936-261-1561; Email: [pvtutoring@pvamu.edu](mailto:pvtutoring@pvamu.edu); [University Tutoring Website](#)

## Writing Center

The Writing Center provides well-trained peer tutors to assist students with writing assignments at any stage of the writing process. Tutors help students with various writing tasks from understanding assignments, brainstorming, drafting, revising, editing, researching, and integrating sources. Students have free access to Grammarly online writing assistance. Grammarly is an automated proofreading and plagiarism detection tool. Students must register for Grammarly by using their student email address. In addition, students have access to face-to-face and virtual tutoring services either asynchronously via email or synchronously via Zoom. Location: J. B. Coleman Library, Rm. 209; Phone: 936-261-3724; [Writing Center Website](#), [Grammarly Registration](#)

## Panther Navigate

Panther Navigate is a proactive system of communication and collaboration between faculty, academic advisors, and students that is designed to support student success by promptly identifying issues and allowing for intervention. Panther Navigate helps students by providing a central location to schedule advising appointments, view campus resources, and request assistance. Students who recognize that they have a problem that negatively affects their academic performance or ability to continue school may self-refer an academic early alert. To do so, students will log in to Canvas and click on Student Alerts on the left sidebar within a course. Students also have the option to download the Navigate Student app. Phone: 936-261-5902; [Panther Navigate Website](#)

## Student Counseling Services

The Student Counseling Services offers a range of services and programs to assist students in maximizing their potential for success: short-term individual, couples, and group counseling, as well as crisis intervention, outreach, consultation, and referral services. The staff is licensed by the State of Texas and assists students who are dealing with academic skills concerns, situational crises, adjustment problems, and emotional difficulties. Information shared with the staff is treated confidentially and in accordance with Texas State Law. Location: Hobart Taylor, 2<sup>nd</sup> floor; Phone: 936-261-3564; [Health & Counseling Center Website](#)

### Office of Testing Services

The Office of Testing Services serves to facilitate and protect the administration of educational and professional exams to aid students, faculty, staff, and the community in their academic and career goals. We provide proctoring services for individuals who need to take exams for distance or correspondence courses for another institution, exams for independent study courses, or make-up exams. In order for a proctored exam to be administered by our office, the instructor of the course must first submit the online PVAMU Testing Services – Test Proctoring Form (this form can only be completed by the instructor) to the Office of Testing Services 72 hours prior to the first exam being administered. Once the Test Proctoring Form has been submitted, the instructor will inform their testers so they can then register for an appointment with our office on one of the selected proctored exam test dates within the testing window for the exam and pay the applicable fees. To access the OTS – Test Proctoring Form, to schedule a proctored exam appointment, or to find more information about our proctoring services, please visit the [OTS – Proctoring Service website](#). Location: Wilhelmina Delco, 3<sup>rd</sup> Floor, Rm. 305; Phone: 936-261-3627; Email: [aetesting@pvamu.edu](mailto:aetesting@pvamu.edu); [Testing Website](#)

### Office of Diagnostic Testing and Disability Services

The Americans with Disabilities Act (ADA) is a federal anti-discrimination statute that provides comprehensive civil rights protection for persons with disabilities. Among other things, this legislation requires that all students with disabilities be guaranteed a learning environment that provides for reasonable accommodation of their disabilities. If you believe you have a disability requiring an accommodation, contact the Office of Disability Services. As a federally-mandated educational support unit, the Office of Disability Services serves as the repository for confidential disability files for faculty, staff, and students. For persons with a disability, the Office develops individualized ADA letters of request for accommodations. Other services include learning style inventories, awareness workshops, accessibility pathways, webinars, computer laboratory with adapted hard and software, adapted furniture, proctoring non-standardized test administrations, ASL interpreters, ALDs, digital recorders, Livescribe, and a comprehensive referral network across campus and the broader community. Location: Hobart Taylor, Rm. 1D128; Phone: 936-261-3583; [Disability Services Website](#)

### Center for Instructional Innovation and Technology Services (CIITS)

Distance Learning, also referred to as Distance Education, is the employment of alternative instructional delivery methods to extend programs and services to persons unable to attend classes in the traditional manner. CIITS supports student learning through online, hybrid, web-assist, and 2-way video course delivery. For more details and contact information, visit [CIITS Student Website](#). Phone: 936-261-3283 or email: [ciits@pvamu.edu](mailto:ciits@pvamu.edu).

### Veteran Affairs

Veteran Services works with student veterans, current military, and military dependents to support their transition to the college environment and continued persistence to graduation. The Office coordinates and certifies benefits for both the G.I. Bill and the Texas Hazlewood Act. Location: Evans Hall, Rm. 102; Phone: 936-261-3563; [Veteran Affairs Website](#)

### Office for Student Engagement

The Office for Student Engagement delivers comprehensive programs and services designed to meet the co-curricular needs of students. The Office implements inclusive and accessible programs and services that enhance student development through exposure to and participation in diverse and relevant social, cultural, intellectual, recreational, community service, leadership development, and campus governance. Location: Memorial Student Center, Rm. 221; Phone: 936-261-1340; [Student Engagement Website](#)

### Center for Careers & Professional Development

This center supports students through professional development, career readiness, and placement and employment assistance. The center provides one-on-one career coaching, interview preparation, resume and letter writing, and career exploration workshops and seminars. Services are provided for students at the Northwest Houston Center and College of Nursing in the Medical Center twice a month or on a requested basis. Distance Learning students are encouraged to visit the center website for information regarding services provided. Location: Anderson Hall, 2<sup>nd</sup> floor; Phone: 936-261-3570; [Center for Careers & Professional Development Website](#)

## University Rules and Procedures

### Academic Misconduct

Academic dishonesty is defined as any form of cheating or dishonesty that has the effect or intent of interfering with any academic exercise or fair evaluation of a student's performance. The college faculty can provide additional information, particularly related to a specific course, laboratory, or assignment.

You are expected to practice academic honesty in every aspect of this course and all other courses. Make sure you are familiar with the *University Administrative Guidelines on Academic Integrity*, which can be found on the [Academic Integrity webpage](#). Students who engage in academic misconduct are subject to university disciplinary procedures. As listed in the *University Administrative Guidelines on Academic Integrity*, the University Online Catalog, and the Student Code of Conduct, the following are examples of prohibited conduct. This list is not designed to be all-inclusive or exhaustive. In addition to academic sanctions, any student found to have committed academic misconduct that is also a violation of criminal law may also be subject to disciplinary review and action by the Office of Student Conduct (as outlined in the Student Code of Conduct).

### Forms of Academic Dishonesty:

1. **Cheating:** Deception in which a student misrepresents that he/she has mastered information on an academic exercise that he/she has not learned, giving or receiving aid unauthorized by the instructor on assignments or examinations. Examples: unauthorized use of notes for a test; using a "cheat sheet" on a quiz or exam; any alteration made on a graded test or exam which is then resubmitted to the teacher;
2. **Plagiarism:** Careless or deliberate use of the work or the ideas of another; representation of another's work, words, ideas, or data as your own without permission or appropriate acknowledgment. Examples: copying another's paper or answers, failure to identify information or essays from the internet and submitting or representing it as your own; submitting an assignment which has been partially or wholly done by another and claiming it as yours; not properly acknowledging a source which has been summarized or paraphrased in your work; failure to acknowledge the use of another's words with quotation marks;
3. **Collusion:** When more than one student or person contributes to a piece of work that is submitted as the work of an individual;
4. **Conspiracy:** Agreeing with one or more persons to commit an act of academic/scholastic dishonesty; and
5. **Multiple Submission:** Submission of work from one course to satisfy a requirement in another course without explicit permission. Example: using a paper prepared and graded for credit in one course to fulfill a requirement and receive credit in a different course.

### PVAMU's General Statement on the Use of Generative Artificial Intelligence Tools in the Classroom

Generative Artificial Intelligence (GAI), specifically foundational models that can create writing, computer code, and/or images using minimal human prompting, are increasingly becoming pervasive. Even though ChatGPT is one of the most well-known GAIs currently available, this statement includes any and all past, current, and future generations of GAI software. Prairie View A&M University expects that all work produced for a grade in any course, be it face-to-face or virtual, will be the sole product of a student's endeavors to meet those academic goals. However, should an instructor permit their students to use artificial intelligence as a resource or tool, students must not substitute the substance of their original work with the results of using such GAI tools. This clearly violates the [University's Administrative Guidelines on Academic Integrity](#) and its underlying academic values.

### Nonacademic Misconduct

The university respects the rights of instructors to teach and students to learn. Maintenance of these rights requires campus conditions that do not impede their exercise. Campus behavior that interferes with either (1) the instructor's ability to conduct the class, (2) the ability of other students to profit from the instructional program, or (3) campus behavior that interferes with the rights of others will not be tolerated. An individual engaging in such



disruptive behavior may be subject to disciplinary action. The Office of Student Conduct will adjudicate such incidents under nonacademic procedures.

### Sexual Misconduct

Sexual harassment of students and employees at Prairie View A&M University is unacceptable and will not be tolerated. Any member of the university community violating the university's sexual harassment policy will be subject to disciplinary action. In accordance with the Texas A&M University System guidelines, your instructor is obligated to report to the Office of Title IX Compliance ([titleixteam@pvamu.edu](mailto:titleixteam@pvamu.edu)) any instance of sexual misconduct involving a student, which includes sexual assault, stalking, dating violence, domestic violence, and sexual harassment, about which the instructor becomes aware during this course through writing, discussion, or personal disclosure. The faculty and staff of PVAMU actively strive to provide a learning, working, and living environment that promotes respect that is free from sexual misconduct, discrimination, and all forms of violence. If students, faculty, or staff would like assistance or have questions, they may contact the Title IX Coordinator, Dr. Zakiya Brown, at 936-261-2144 or [titleixteam@pvamu.edu](mailto:titleixteam@pvamu.edu). More information can be found at [Title XI Website](#), including confidential resources available on campus.

### Protections and Accommodations for Pregnant and Parenting Students

The U.S. Department of Education's Office for Civil Rights (OCR) enforces, among other statutes, Title IX of the Education Amendments of 1972. Title IX protects people from discrimination based on sex, sexual orientation, and gender identity in education programs or activities that receive federal financial assistance. This protection includes those who may be pregnant and parenting. Title IX states: "No person in the United States shall, on the basis of sex, be excluded from participation in, be denied the benefits of, or be subjected to discrimination under any education program or activity receiving Federal financial assistance." Students seeking accommodations related to pregnancy or parenting should contact the Office of Title IX for information, resources, and support at [titleixteam@pvamu.edu](mailto:titleixteam@pvamu.edu). Additional information and/or support may be provided by the Office of Disability Services or the Office of the Dean of Students.

### Non-Discrimination Statement

Prairie View A&M University does not discriminate on the basis of race, color, sex, religion, national origin, age, disability, genetic information, veteran status, sexual orientation, or gender identity in its programs and activities. The University is committed to supporting students and complying with The Texas A&M University System non-discrimination policy. It seeks to establish an environment that is free of bias, discrimination, and harassment. If you experience an incident of discrimination or harassment, we encourage you to report it. If you would like to speak with someone who may be able to afford you privacy or confidentiality, there are individuals who can meet with you. The Director of Equal Opportunity & Diversity has been designated to handle inquiries regarding the non-discrimination policies and can be reached at Harrington Science Building, Suite 109, or by phone at 936-261-1744 or 1792.

### Class Attendance Policy (See the University Online Catalog for Full Attendance Policy)

Prairie View A&M University requires regular class attendance. Attending all classes supports the full academic development of each learner, whether classes are taught with the instructor physically present or via distance learning technologies such as interactive video and/or the Internet. Excessive absenteeism, whether excused or unexcused, may result in a student's course grade being reduced or in the assignment of a grade of "F." Absences are accumulated beginning with the first day of class during regular semesters and summer terms. Each faculty member will include the University's attendance policy in each course syllabus.

### Makeup Work for Legitimate Absences

Prairie View A&M University recognizes that there are a variety of legitimate circumstances in which students will miss coursework and that accommodations for makeup work will be made. If a student's absence is **excused**, the instructor must either provide the student an opportunity to make up any quiz, exam, or other work contributing to the final grade or provide a satisfactory alternative by a date agreed upon by the student and instructor. Students are encouraged to work with instructors to complete makeup work before known scheduled absences (University-sponsored events, administrative proceedings, etc.). Students are responsible for planning their schedules to avoid excessive conflicts with course requirements.

### Absence Verification Process

All non-athletic absences (e.g., Medical, Death/Funeral, Court/Legal-related, etc.) for which a student seeks to obtain a valid excuse must be submitted to the Dean of Students/Office of Student Conduct, with supporting documentation, for review and verification. Please use the [Online Reporting Forms](#) to access/complete/submit the *Request for a University Excused Absence* form for an excuse. Upon receipt, a staff member will verify the documentation and provide an official university excuse, if applicable. The student is responsible for providing the official university excuse issued by the Office for Student Conduct to the professor(s). Questions should be directed to the Dean of Students via email: [deanofstudents@pvamu.edu](mailto:deanofstudents@pvamu.edu) or phone: (936) 261-3550 or Office for Student Conduct via email: [studentconduct@pvamu.edu](mailto:studentconduct@pvamu.edu) or phone: (936) 261-3524.

#### Student Academic Appeals Process

Authority and responsibility for assigning grades to students rest with the faculty. However, in those instances where students believe that miscommunication, errors, or unfairness of any kind may have adversely affected the instructor's assessment of their academic performance, the student has a right to appeal by the procedure listed in the University Online Catalog and by doing so within thirty days of receiving the grade or experiencing any other problematic academic event that prompted the complaint.

#### Technical Considerations

##### **Minimum Recommended Hardware and Software:**

- Intel PC or laptop with Windows 10 or later version; Mac with OS Catalina
- Smartphone or iPad/tablet with wi-fi\*
- High-speed internet access
- 8 GB memory
- Hard drive with 320 GB storage space
- 15" monitor, 1024 x 768, color
- Speakers (internal or external)
- Microphone and recording software
- Keyboard & mouse
- Most current version of Google Chrome, Safari, or Firefox

**Note:** Be sure to enable Java & pop-ups in the web browser preferences

\* Some courses may require remote proctoring. At this time only Chromebooks, laptops, and desktops running Windows or Mac work with our proctoring solution, but iPads are not compatible. Most other applications will work with Android or Apple tablets and smartphones.

##### **Participants should have a basic proficiency of the following computer skills:**

- Sending and receiving email
- A working knowledge of the Internet
- Microsoft Word (or a program convertible to Word)
- Acrobat PDF Reader
- Windows or Mac OS
- Video conferencing software (Zoom)

#### *Netiquette* (online etiquette)

Students are expected to participate in all discussions and virtual classroom chats as directed. Students are to be respectful and courteous to others on discussion boards. Foul or abusive language will not be tolerated. Do not use ALL CAPS for communicating to others AS IT CAN BE INTERPRETED AS YELLING. Avoid slang terms such as "wassup?" and texting abbreviations such as "u" instead of "you." Limit and possibly avoid the use of emoticons. Be cautious when using humor or sarcasm as tone is sometimes lost in an email or discussion post, and the message might be taken seriously or sound offensive.

#### Video Conferencing Etiquette

When using Zoom, WebEx, or other video conferencing tools, confirm the visible area is tidy, clear of background clutter, inappropriate or offensive posters, and other distractions. Ensure you dress appropriately and avoid using high traffic or noisy areas. Stay muted when you are not speaking and avoid eating/drinking during the session.

Before the class session begins, test audio, video, and lighting to alleviate technology issues.

#### Technical Support

Students should go to [Password Reset Tool](#) if they have password issues. The page will provide instructions for resetting passwords and contact information if login issues persist. For other technical questions regarding eCourses, call the Center for Instructional Innovation and Technology Services at 936-261-3283 or email [ciits@pvamu.edu](mailto:ciits@pvamu.edu).

#### Communication Expectations and Standards

Emails or discussion postings will receive a response from the instructor, usually in less than 48 hours. Urgent emails should be marked as such. Check regularly for responses.

#### Discussion Requirement

Online courses often require minimal to no face-to-face meetings. However, conversations about the readings, lectures, materials, and other aspects of the course can occur in a seminar fashion. The use of the discussion board will accomplish this. The instructor will determine the exact use of discussion boards.

**It is strongly suggested** that students type their discussion postings in a word processing application such as Word and save it to their PC or a removable drive before posting to the discussion board. This is important for two reasons: 1) If for some reason your discussion responses are lost in your online course, you will have another copy; 2) Grammatical errors can be greatly minimized by the use of the spell-and-grammar check functions in word processing applications. Once the post(s) have been typed and corrected in the word processing application, copy and paste to the discussion board.

#### COVID-19 Campus Safety Measures

In accordance with the latest guidelines from the PVAMU Health Services, the following measures are in effect until further notice.

- Students who are ill will be asked to adhere to best practices in public health, such as masking, handwashing, and social distancing, to help reduce the spread of illness across campus.
- Mandatory self-reporting will no longer be required by students. Students will be responsible for communicating with their professors regarding COVID, similarly to any other illness.
- There will be no mandatory isolation. Students who are too ill to engage in classroom activities will be responsible for securing the appropriate documentation to support the absence.
- Students who self-isolate will be responsible for communicating with their professors and securing an excuse from Student Conduct.
- All students will have access to [TimelyCare](#), a telehealth platform that provides virtual medical care 24/7 and by appointment in the Student Health Clinic. Students are encouraged to enroll with TimelyCare at the beginning of the semester, at [timelycare.com/pvamu](http://timelycare.com/pvamu).
- Students will have access to COVID testing in the Student Health Clinic by appointment. Testing is for students who are symptomatic ONLY.

# ATTACHMENT 1 Proctorio Setting

[Details](#) [Questions](#) [Mastery Paths](#) **Proctorio Settings**

## ▼ Saved Profiles





You have not saved any profiles yet :(

Saving Proctorio settings profiles allows you to apply your favorite exam settings with a single click.

## ▼ Proctorio Exam Settings











Exam settings can not be changed once the first test taker has started the exam.

### ▼ Recording Options




-  Record Video
-  Record Audio
-  Record Screen
-  Record Web Traffic

Record Screen will capture the test taker's desktop for the entire exam, allowing you to see exactly what they were seeing on screen.



### ▼ Lock Down Options

-  Severe (0s)
-  Only One Screen
-  No New Tabs
-  Close Open Tabs
-  Disable Printing
-  Disable Clipboard
-  Block Downloads
-  Clear Cache
-  Disable Right Click
-  Prevent Re-entry

### ▼ Verification Options

-  Verify Video
-  Verify Audio
-  Verify Desktop

### ▼ In-Quiz Tools

-  Scientific Calculator
-  Basic Whiteboard

ATTACHMENT 2 - **How to write the powers and units in short answers on canvas**

**Powers for scientific notation**

$$1020000 = 1.02E6$$

$$0.0021 = 2.1E-3$$

The number of decimals to round an answer will be indicated in the question.

**Units in canvas:**

**Time:**

Days = days

Seconds = s

Minutes = min

Hours = hr

**Chemical units:**

Mol = mol

Molarity = M

**Temperature:**

Kelvin = K

Fahrenheit = °F

Celsius = °C

**Energy:**

Joules = J

Kilo Joules = kJ,

**Mass:**

Grams = g

**Pressure:**

Atmospheres = atm

**Volume:**

Milliliters = mL

Liters = L

Density = d

Meters = m

**Combine and Reverse units** (do not write any of these symbols =, x, •, \*, ^) leave an space between 2 units (example, g m); use / to indicate reverse (example, g/mol)

1/unit (Example, 1/s, 1/K)

Unit - unit (Example, liters atmospheres = L atm)

Unit/unit - unit (Example, Joules per mol per kelvin = J/mol K)

Any other specific unit notation will be indicated in the respective exam